

**Long He<sup>a\*</sup> and Seik Weng Ng<sup>b</sup>**

<sup>a</sup>College of Chemistry and Chemical Engineering, China West Normal University, Nanchong 637002, People's Republic of China, and <sup>b</sup>Department of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia

Correspondence e-mail: cwnuchem@163.com

**Key indicators**

Single-crystal X-ray study  
 T = 153 K  
 Mean  $\sigma(\text{C}-\text{C}) = 0.002 \text{ \AA}$   
 R factor = 0.019  
 wR factor = 0.055  
 Data-to-parameter ratio = 15.2

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

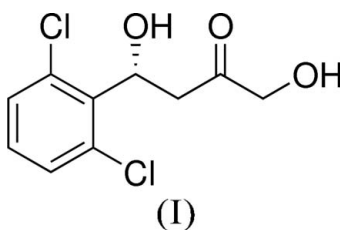
**4-(2,6-Dichlorophenyl)-1,4-dihydroxybutan-2-one**

Molecules of the title compound,  $\text{C}_{10}\text{H}_{10}\text{Cl}_2\text{O}_3$ , are linked by  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bonds into a sheet motif.

Received 22 November 2006  
 Accepted 24 November 2006

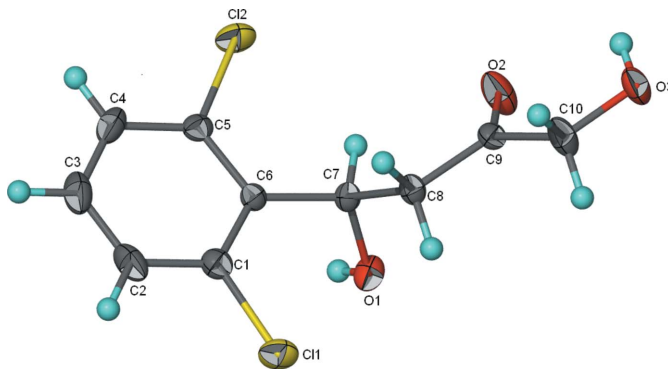
**Comment**

The title compound, (I), is an important reagent for the synthesis of 1,2,4-triols and substituted tetrahydrofurans (Chen *et al.*, 1987; Zheng *et al.*, 2005). It crystallizes as a monomeric molecule that interacts with adjacent molecules through hydrogen bonds (Table 1) to form a sheet motif. The carbonyl group is not involved in such interactions.



**Experimental**

To a solution of 2,6-dichlorobenzaldehyde (87.5 mg, 0.5 mmol) and (*S*)-*N*-phenylpyrrolidine-2-carboxamide (19 mg, 0.5 mmol) in a mixture of water (1 ml) and tetrahydrofuran (1 ml) was added hydroxyacetone (0.5 ml). The mixture was stirred at 273 K for 5 d. It was then treated with saturated ammonium chloride solution. The aqueous layer was extracted with ethyl acetate; the extract was dried over anhydrous magnesium sulfate. The solvent was removed and then purified by column chromatography on silica gel. Crystals were obtained by using ethanol as solvent.



**Figure 1**  
 The molecular structure of (I) with 70% probability displacement ellipsoids (arbitrary spheres for H atoms).

## Crystal data

$C_{10}H_{10}Cl_2O_3$	$Z = 2$
$M_r = 249.08$	$D_x = 1.566 \text{ Mg m}^{-3}$
Monoclinic, $P2_1$	Mo $K\alpha$ radiation
$a = 7.7674 (2) \text{ \AA}$	$\mu = 0.60 \text{ mm}^{-1}$
$b = 8.7852 (2) \text{ \AA}$	$T = 153 (2) \text{ K}$
$c = 8.3555 (3) \text{ \AA}$	Block, colorless
$\beta = 112.093 (1)^\circ$	$0.45 \times 0.42 \times 0.36 \text{ mm}$
$V = 528.30 (3) \text{ \AA}^3$	

## Data collection

Rigaku R-AXIS RAPID diffractometer	5238 measured reflections
$\omega$ scans	2205 independent reflections
Absorption correction: multi-scan ( <i>ABSCOR</i> ; Higashi, 1995)	2189 reflections with $I > 2\sigma(I)$
$T_{\min} = 0.775$ , $T_{\max} = 0.814$	$R_{\text{int}} = 0.013$
	$\theta_{\max} = 27.5^\circ$

## Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0395P)^2 + 0.0574P]$
$R[F^2 > 2\sigma(F^2)] = 0.019$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.055$	$(\Delta/\sigma)_{\max} = 0.001$
$S = 1.08$	$\Delta\rho_{\max} = 0.30 \text{ e \AA}^{-3}$
2205 reflections	$\Delta\rho_{\min} = -0.20 \text{ e \AA}^{-3}$
145 parameters	Extinction correction: <i>SHELXL97</i>
H atoms treated by a mixture of independent and constrained refinement	Extinction coefficient: 0.145 (8)
	Absolute structure: Flack (1983), 913 Friedel pairs
	Flack parameter: 0.01 (4)

Table 1

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$O1-H1O\cdots O3^i$	0.85 (1)	2.02 (1)	2.844 (1)	164 (2)
$O3-H3O\cdots O1^{ii}$	0.85 (1)	2.24 (2)	3.032 (2)	156 (2)

Symmetry codes: (i)  $x - 1, y, z$ ; (ii)  $-x + 2, y + \frac{1}{2}, -z$ .

Carbon-bound H atom were positioned geometrically ( $C-H = 0.95-1.00 \text{ \AA}$ ) and were allowed to ride on the parent C atoms, with  $U_{\text{iso}}(H) = 1.2U_{\text{eq}}(C)$ . The hydroxy H atoms were located in a difference Fourier map and were refined with a distance restraint of  $O-H = 0.85 (1) \text{ \AA}$ ; the displacement parameters were freely refined.

Data collection: *RAPID-AUTO* (Rigaku, 2004); cell refinement: *RAPID-AUTO*; data reduction: *RAPID-AUTO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *pubCIF* (Westrip, 2006).

The diffraction measurements were made at the Centre for Testing and Analysis, Chengdu Branch, Chinese Academy of Sciences. We thank China West Normal University (05B022) and the University of Malaya for supporting this study.

## References

- Barbour, L. J. (2001). *J. Supramol. Chem.* **1**, 189–191.  
 Chen, K. M., Hardtman, G. E., Prasad, K. & Repic, O. (1987). *Tetrahedron Lett.* **28**, 155–158.  
 Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.  
 Higashi, T. (1995). *ABSCOR*. Rigaku Corporation, Tokyo.  
 Rigaku (2004). *RAPID-AUTO*. Rigaku/MSC Inc., The Woodlands, Texas, USA.  
 Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.  
 Westrip, S. P. (2006). *pubCIF*. In preparation.  
 Zheng, T., Narayan, R. S., Schomaker, J. M. & Borhan, B. (2005). *J. Am. Chem. Soc.* **127**, 6946–6947.